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## Synthesis of *N*-Aryl Uracils and Hypoxanthines and Their Biological Properties

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## SYNTHESIS OF *N*-ARYL URACILS AND HYPOXANTHINES AND THEIR BIOLOGICAL PROPERTIES

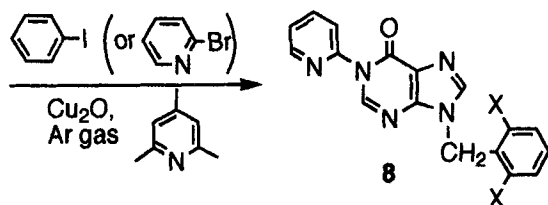
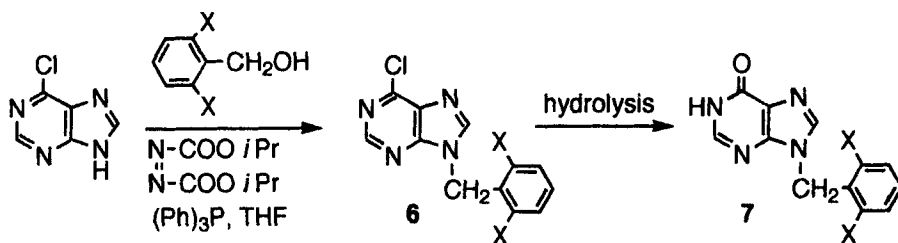
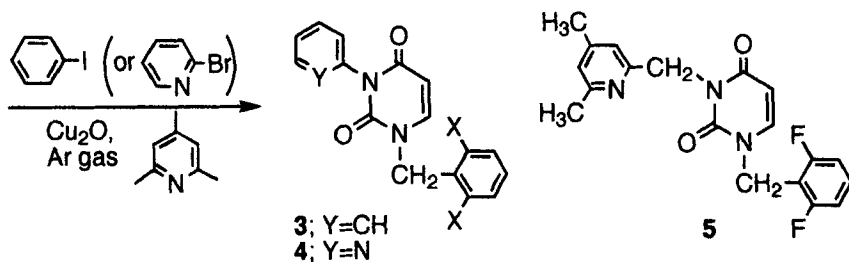
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**ABSTRACT:** 1-Benzyluracils **2a,b** were treated with iodobenzene in the presence of cuprous oxide in 2,4,6-trimethylpyridine at 180°C to give the *N*<sup>1</sup>-phenyl derivatives **3a** and **3b** in 47% and 55%, respectively. Similar reaction of **2a** with 2-bromopyridine at 120°C gave the 3-(2-pyridinyl)uracil **4a** in 42% yield. However, unusual product **5** as well as 3-(2-pyridinyl) derivative **4b** were obtained in the case of **2b**. The structure of **5** was identified as 1-(2,6-difluorobenzyl)-3-[(2,4-dimethyl-2-pyridinyl)methyl]uracil from spectroscopic data. Reaction of the hypoxanthines **7a,b** with 2-bromopyridine gave the 1-(2-pyridinyl)hypoxanthines **8a,b** in low yields. But *N*-phenylation of **7a,b** were unsuccessful.

### INTRODUCTION

Since treatment of epilepsy by drugs fail to control seizure satisfactory and experience significant side effects in many case, the need for improved drug is increasing.<sup>1,2)</sup> Kelley *et al.* reported the syntheses and anticonvulsant activities of 9-(2-fluorobenzyl)-6-methyladenine (BW A78U)<sup>3)</sup> and its carbon-nitrogen isostere (534U87).<sup>4)</sup> Recently, uridine was identified as a sleep-promoting substances<sup>5)</sup> and *N*<sup>3</sup>-benzyluridine was found to have CNS-depressant effects.<sup>6)</sup> These backgrounds prompted us to investigate the synthesis of the *N*-aryl derivatives of 1-benzyluracils and hypoxanthines.



## RESULTS AND DISCUSSION

1-Benzyluracils **2a,b** were prepared from 2,4-dimethoxypyrimidine via 4-methoxy-2(1*H*)-pyrimidinones **1a,b** in a similar manner for the synthesis of 1-methyluracil.<sup>9)</sup> Thus reaction of **2a** with iodobenzene in the presence of cuprous oxide in 2,4,6-trimethylpyridine

at 180 °C gave **3a** in 47%. Compound **3a** showed similar UV spectrum to that of 3-phenyluridine, suggesting *N*<sup>3</sup>-phenyluracil structure. Analytical data also supported the structure of **3a**. This reaction was applied to the 2,6-difluorobenzyluracil **2b**. 3-Phenyl derivative **3b** was obtained in 55% yield and recovered starting material in 9.1%. Compound **2a** was also treated with 2-bromopyridine at 120 °C using the same catalyst and solvent to give the 3-(2-pyridinyl)uracil **4a** in 42% yield. Similar reaction of **2b** with 2-bromopyridine was carried out to afford the desired product **4b** and unusual product **5** in 52% and 14.6%, respectively. The UV spectrum suggested that compound **5** is *N*<sup>3</sup>-alkylated product. The <sup>1</sup>H-NMR spectrum revealed appearance of two singlet at 2.20 and 2.42 ppm attributable to methyl protons. Signals of two methylene protons were also observed at 5.05 and 5.20 ppm, indicating 1-(2,6-difluorobenzyl)-3-[(2,4-dimethyl-6-pyridinyl)methyl] uracil structure. Mass spectroscopic data (*M*<sup>+</sup>, *m/z* 251) also supported the structure. The reaction mechanism to form **5** is under investigation.

9-Benzylhypoxanthine **7a** was prepared by hydrolysis of 9-benzyl-6-chloropurine **6a**.<sup>10,11)</sup> 9-(2,6-Difluorobenzyl)hypoxanthine **7b** was also prepared by hydrolysis of 6-chloro-9-(2,6-difluorobenzyl)purine **6b**. The hypoxanthines **7a,b** were treated with 2-bromopyridine in a manner similar to that of **4a** to give the 2-pyridinyl derivatives **8a** and **8b** in low yields. But *N*-phenylation of **7a,b** was unsuccessful. In the case of **7b** unknown product was obtained.

Preliminary biological activities of compounds **2b**, **4b**, **6b** and **7b** were evaluated. No significant response was observed in the assay of convulsion by maximal electroshock.<sup>3,4)</sup> However, compound **6b** showed weak inhibition activity against phosphodiesterase I (IC<sub>50</sub> 18 μg/ml).

## EXPERIMENTAL

Melting points (mp) were determined using a Yanagimoto micro-melting point

apparatus (hot stage type) and are uncorrected. UV spectra were recorded with a Shimadzu UV-190 digital spectrometer. Low resolution mass spectra were obtained on a Shimadzu-LKB 9000B mass spectrometer in the direct-inlet mode. High resolution mass spectra were obtained on a JMS AX-500 spectrometer in the direct-inlet mode.  $^1\text{H}$ -NMR spectra were recorded on either Varian UNITY 200 (200 MHz) or Varian UNITY 600 (600 MHz) in  $\text{CDCl}_3$  (or dimethyl sulfoxide ( $\text{DMSO}$ )- $d_6$ ) with tetramethylsilane as an internal standard. Merck Art 5554 plates precoated with silica gel 60 containing fluorescent indicator  $\text{F}_{254}$  were used for thin-layer chromatography and silica gel 60 (Merck 7734, 60 - 200 mesh) was employed for column chromatography.

**1-Benzyl-4-methoxy-2(1*H*)-pyrimidinone (1a).** A mixture of 2,4-dimethoxy-pyrimidine (4.22 g, 30.1 mmol) and benzyl bromide (36 ml, 10 eq.) was kept at 40 °C overnight. The crystals were collected by filtration and the filtrate was applied to the column of silica gel G ( $3.3 \times 30$  cm). The column was washed with benzene (1 l) and eluted with  $\text{CHCl}_3$  (1 l) and the fraction of the product was concentrated to afford another crystals. Yield 4.56 g (70%). mp 113-117 °C. UV  $\lambda$  max (MeOH) 277 nm. MS  $m/z$ : 216 ( $\text{M}^+$ ), 201 ( $\text{M}^+ - \text{CH}_3$ ).  $^1\text{H}$ -NMR ( $\text{DMSO}-d_6$ )  $\delta$  : 7.2-7.5 (6H, m, H6,  $\text{C}_6\text{H}_5$ ), 5.85 (1H, d,  $J = 7.3$  Hz, H5), 5.06 (2H, s,  $-\text{CH}_2-$ ), 3.97 (3H, s,  $\text{OCH}_3$ ).

**1-Benzyluracil (2a).** To a suspension of **1a** (2.17 g, 10 mmol) in 1,4-dioxane (50 ml) was added concentrated HCl and the solution was heated under reflux for 1 h, then cooled. Recrystallization of the product from EtOH gave white crystals (1.33 g, 66%). mp 169-171 °C (lit.<sup>12</sup> 175 °C). Anal. Calcd for  $\text{C}_{11}\text{H}_{10}\text{N}_2\text{O}_2$ : C, 65.34; H, 4.98; N, 13.85. Found: C, 65.08; H, 4.98; N, 13.61. MS  $m/z$ : 202 ( $\text{M}^+$ ). UV  $\lambda$  max (MeOH) 265 nm.  $^1\text{H}$ -NMR ( $\text{DMSO}-d_6$ )  $\delta$  : 11.33 (1H, br s,  $\text{N}^3\text{-H}$ ), 7.76 (1H, d,  $J = 7.88$  Hz, H6), 7.25 - 7.42 (5H, m, Ph), 5.60 (1H, d,  $J = 7.69$  Hz, H5), 4.88 (2H, s,  $-\text{CH}_2-$ ).

**1-(2,6-Difluorobenzyl)-4-methoxy-2(1H)-pyrimidinone (1b).** To a solution of 2,4-dimethoxypyrimidine (1.32 g, 10 mmol) in dry THF (10 ml) was added 2,6-difluorobenzyl bromide (2.07 g, 10 mmol) and the solution was heated at 80°C for 1 h. After cooling, the solution was diluted with AcOEt (10 ml) and chromatographed over a column of silica gel G (3.0×45 cm) using a gradient of hexane-AcOEt = 1:1 and AcOEt as a eluant. Evaporation of the fraction and crystallization from AcOEt gave white crystals (1.08 g, 42%). mp 143-145°C. *Anal.* Calcd for C<sub>12</sub>H<sub>10</sub>F<sub>2</sub>N<sub>2</sub>O<sub>2</sub>: C, 57.15; H, 4.00; N, 11.11. Found: C, 57.15; H, 3.98; N, 11.01. MS *m/z*: 252 (M<sup>+</sup>). UV  $\lambda$  max (MeOH) 274 nm. <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 7.40 (1H, dt, *J* = 7.33, 1.10 Hz, H6), 7.30 (1H, m, H4'), 6.94 (2H, m, H3', H5'), 5.84 (1H, d, *J* = 7.33 Hz, H5), 5.12 (2H, s, -CH<sub>2</sub>-), 3.94 (3H, s, OCH<sub>3</sub>).

**1-(2,6-Difluorobenzyl)uracil (2b).** A solution of **1b** (252 mg, 1 mmol) in a mixture of 1,4-dioxane (50 ml) and concentrated HCl (0.1 ml) was heated under reflux for 1 h, then cooled. Concentration of the solution afforded solid, which was recrystallized from EtOH to give white crystals (190 mg, 79%). mp 226-227°C. *Anal.* Calcd for C<sub>11</sub>H<sub>8</sub>F<sub>2</sub>N<sub>2</sub>O<sub>2</sub>: C, 55.47; H, 3.39; N, 11.76. Found: C, 55.43; H, 3.32; N, 11.59. MS *m/z*: 238 (M<sup>+</sup>). UV  $\lambda$  max (MeOH) 263 nm. <sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>)  $\delta$ : 11.27 (1H, br s, N<sup>3</sup>-H), 7.71 (1H, dt, *J* = 7.88, 1.28 Hz, H6), 7.43 (1H, m, H4'), 7.12 (2H, m, H3', H5'), 5.58 (1H, d, *J* = 8.06 Hz, H5), 4.95 (2H, s, -CH<sub>2</sub>-).

**1-Benzyl-3-phenyluracil (3a).** To a solution of **2a** (404 mg, 2 mmol) and copper(I) oxide (286 mg, 2 mmol) in 2,4,6-trimethylpyridine (5 ml) was added iodobenzene (1.2 ml, 10.7 mmol) and the solution was heated at 180°C under Ar atmosphere for 20 h. Reaction was continued with additional iodobenzene (1.2 ml, 10.7

mmol) for further 20 h and the solution was diluted with  $\text{CH}_2\text{Cl}_2$  (80 ml). The insoluble was removed by filtration and the filtrate was applied to a column of silica gel G, then eluted with 17% AcOEt in hexane and hexane-AcOEt = 1 : 2. The second fraction was evaporated and the residue was crystallized from AcOEt to give white crystals (259.6 mg, 47%). mp 161-162 °C. *Anal.* Calcd for  $\text{C}_{17}\text{H}_{14}\text{N}_2\text{O}_2$ : C, 73.37; H, 5.07; N, 10.07. Found: C, 73.26; H, 4.98; N, 10.01. MS  $m/z$ : 278 ( $\text{M}^+$ ). UV  $\lambda$  max (MeOH) 266 nm.  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 7.21 - 7.55 (11H, m, Ph,  $\text{N}^3$ -Ph, H6), 5.85 (1H, d,  $J$  = 8.06 Hz, H5), 4.95 (2H, s,  $-\text{CH}_2-$ ).

**1-Benzyl-3-(pyridin-2-yl)uracil (4a).** To a solution of **2a** (202 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in 2,4,6-trimethylpyridine (6 ml) was added 2-bromopyridine (0.48 ml, 5 mmol) and the solution was heated at 120 °C under Ar atmosphere for 20 h. Work-up of the solution in a manner similar to that described in the section of **3a** gave a caramel (114 mg, 42%). mp 133-134 °C. *Anal.* Calcd for  $\text{C}_{16}\text{H}_{13}\text{N}_3\text{O}_2$ : C, 68.81; H, 4.69; N, 15.04. Found: C, 68.90; H, 4.64; N, 14.86. High-resolution MS  $m/z$  Calcd for  $\text{C}_{16}\text{H}_{13}\text{N}_3\text{O}_2$  ( $\text{M}^+$ ) 279.1008. Found: 279.0991. UV  $\lambda$  max (MeOH) 264 nm.  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 8.71 (1H, m, Py 1H), 7.89 (1H, dt,  $J$  = 1.65, 7.51 Hz, Py 1H), 7.23 - 7.44 (8H, m, Ph, Py 2H, H6), 5.85 (1H, d,  $J$  = 7.69 Hz, H5), 4.95 (2H, s,  $-\text{CH}_2-$ ).

**1-(2,6-Difluorobenzyl)-3-phenyluracil (3b).** A solution of **2b** (238 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in the mixture of 2,4,6-trimethylpyridine (5 ml) and iodobenzene (0.6 ml, 5.4 mmol) was heated at 180 °C under Ar atmosphere for 20 h. The resulting solution was worked up in a manner similar to that described in the section of **3a** to give white crystals (176 mg, 55%). mp 183-184 °C. *Anal.* Calcd for

$C_{17}H_{12}F_2N_2O_2$ : C, 64.97; H, 3.85; N, 8.91. Found: C, 64.78; H, 3.85; N, 8.83. MS  $m/z$ : 314 ( $M^+$ ). UV  $\lambda$  max (MeOH) 264 nm.  $^1H$ -NMR ( $CDCl_3$ )  $\delta$ : 7.31–7.52 (7H, m, Ph, H4', H6), 6.92–7.28 (2H, m, H3', H5'), 5.86 (1H, d,  $J$  = 8.06 Hz, H5), 5.04 (2H, s,  $-CH_2-$ ). Evaporation of second fraction gave starting material **2b** (22.1 mg, 9.1%).

**Reaction of 1-(2,6-difluorobenzyl)uracil 2b with 2-bromopyridine.** A solution of **2b** (238 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in the mixture of 2,4,6-trimethylpyridine (5 ml) and 2-bromopyridine (0.48 ml, 5 mmol) was heated at 130 °C under Ar atmosphere for 20 h. The resulting solution was worked up in a manner similar to that described in the section of **3a** to afford three compounds. First fraction: starting material (white crystals, 11 mg, 4.5%). Second fraction: 1-(2,6-difluorobenzyl)-3-[(2,4-dimethyl-2-pyridinyl)methyl]uracil **5** (a caramel, 46.9 mg, 15%), MS  $m/z$ : 357 ( $M^+$ ), High-resolution MS  $m/z$  Calcd for  $C_{19}H_{17}F_2N_3O_2$  ( $M^+$ ) 357.1290. Found: 357.1269. UV  $\lambda$  max (MeOH) 263 nm.  $^1H$ -NMR ( $CDCl_3$ )  $\delta$ : 7.30 (2H, m, H6, H4'), 6.92–7.00 (2H, m, H3', H5'), 6.80 (1H, s, Py H5), 6.64 (1H, s, Py H3), 5.80 (1H, d,  $J$  = 8.06 Hz, H5), 5.20 (2H, s,  $N^3$ - $CH_2$ ), 5.05 (2H, s,  $N^1$ - $CH_2$ ), 2.42 (3H, s,  $CH_3$ ), 2.20 (3H, s,  $CH_3$ ). Third fraction: 1-(2,6-difluorobenzyl)-3-(2-pyridinyl)uracil **4b** (a caramel, 168 mg, 52%), MS  $m/z$ : 315 ( $M^+$ ), High-resolution MS  $m/z$  Calcd for  $C_{16}H_{11}F_2N_3O_2$  ( $M^+$ ) 315.0820. Found: 315.0800. UV  $\lambda$  max (MeOH) 263 nm.  $^1H$ -NMR ( $CDCl_3$ )  $\delta$ : 8.65 (1H, br s, Py 1H), 7.82–7.91 (1H, dt,  $J$  = 2.01, 7.69 Hz, Py 1H), 7.27–7.42 (4H, m, Py 2H, H6, H4'), 6.91–6.99 (2H, m, H3', H5'), 5.86 (1H, d,  $J$  = 8.06 Hz, H5), 5.02 (2H, s,  $-CH_2-$ ).

**9-Benzylhypoxanthine (7a).** A solution of 6-chloro-9-benzylpurine **6a**<sup>10,11</sup> (245 mg, 1 mmol) in 1M HCl (5 ml) was heated under reflux for 1 h and the solution was concentrated to give white crystals, which was recrystallized from EtOH (192 mg, 84%).



mp 284.5-286.5°C. *Anal.* Calcd for  $C_{12}H_{10}N_4O$ : C, 63.66; H, 4.46; N, 24.76. Found: C, 63.56; H, 4.47; N, 24.65. MS  $m/z$ : 226 ( $M^+$ ). UV  $\lambda$  max (MeOH) 250 nm.  $^1H$ -NMR (DMSO- $d_6$ )  $\delta$ : 12.31 (1H, br s,  $N^1$ -H), 8.22 (1H, s, H8), 8.06 (1H, d,  $J=2.47$ , H2), 7.28-7.35 (5H, m, Ph), 5.38 (2H, s,  $-CH_2-$ ).

**Phenylation of 9-benzylhypoxanthine (7a).** A solution of **7a** (226 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in the mixture of 2,4,6-trimethylpyridine (5 ml) and iodobenzene (0.6 ml, 5.4 mmol) was heated at 130°C under Ar atmosphere for 20 h. Spots of **7a** and desired product were not observed on TLC.

**9-Benzyl-1-(pyridin-2-yl)hypoxanthine (8a).** To a solution of **7a** (226 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in 2,4,6-trimethylpyridine (6 ml) was added 2-bromopyridine (0.48 ml, 5 mmol) and the solution was heated at 100°C under Ar atmosphere for 20 h. Reaction was continued for further 44 h at 120°C and work-up of the solution in a manner similar to that described in the section of **3a** gave a solid, which was crystallized from benzene to give white crystals (45 mg, 15%). mp 153°C. *Anal.* Calcd for  $C_{17}H_{13}N_5O$ : C, 67.32; H, 4.32; N, 23.09. Found: C, 67.21; H, 4.35; N, 23.04. MS  $m/z$ : 303 ( $M^+$ ). UV  $\lambda$  max (MeOH) 256 nm.  $^1H$ -NMR ( $CDCl_3$ )  $\delta$ : 8.59-8.63 (1H, m, Py H2), 8.53 (1H, s, H2), 7.88-7.91 (2H, m, Py H4, Py H5), 7.79 (1H, s, H8), 7.27-7.43 (6H, m, Ph, Py H3), 5.38 (2H, s,  $-CH_2-$ ).

**6-Chloro-9-(2,6-difluorobenzyl)purine (6b).** To a solution of 6-chloropurine (154 mg, 1 mmol) was suspended in DMF (10 ml) was added dry  $K_2CO_3$  (138 mg, 1 mmol) and the solution was stirred at 100°C for 30 min. 2,6-Difluorobenzyl bromide (207 mg, 1 mmol) was added to the solution and stirring was continued at 50°C for further 1 h.

After cooling, the insolubles were removed by filtration and the filtrate was evaporated. The residue was chromatographed over a column of silica gel G ( $\phi$  3×40 cm) using a gradient hexane-AcOEt = 1 : 2 and AcOEt to give white crystals (120 mg, 42%). mp 147-148 °C. *Anal.* Calcd for  $C_{12}H_7ClF_2N_4$ : C, 51.35; H, 2.51; N, 19.96. Found: C, 51.40; H, 2.58; N, 20.02. MS  $m/z$ : 280, 282 ( $M^+$ ). UV  $\lambda$  max (MeOH) 264 nm.  $^1H$ -NMR ( $CDCl_3$ )  $\delta$ : 8.80 (1H, s, H2), 8.18 (1H, s, H8), 7.38 (1H, m, H4'), 6.99 (2H, m, H3', H5'), 5.56 (2H, s,  $-CH_2-$ ).

**9-(2,6-Difluorobenzyl)hypoxanthine (7b).** A mixture solution of **6b** (840 mg, 3 mmol) and NaOAc (738 mg, 9 mmol) in AcOH (15 ml) was stirred at 120 °C for 5 h, then cooled. The solution was evaporated to give a residue, to which water (20 ml) was added to give white crystals (694 mg, 88%). mp 302-303 °C. *Anal.* Calcd for  $C_{12}H_8F_2N_4O$ : C, 54.97; H, 3.07; N, 21.37. Found: C, 54.80; H, 3.13; N, 21.24. MS  $m/z$ : 262 ( $M^+$ ). UV  $\lambda$  max (MeOH) 250 nm.  $^1H$ -NMR ( $DMSO-d_6$ )  $\delta$ : 12.29 (1H, br s,  $N^1$ -H), 8.10 (1H, s, H8), 8.03 (1H, s, H2), 7.49 (1H, m, H4'), 7.26 (2H, m, H3', H5'), 5.46 (2H, s,  $-CH_2-$ ).

**Reaction of 9-(2,6-difluorobenzyl)hypoxanthine 7b with iodobenzene.** A solution of **7b** (262 mg, 1 mmol) and copper(I) oxide (143 mg, 1 mmol) in the mixture of 2,4,6-trimethylpyridine (5 ml) and iodobenzene (0.6 ml, 5.36 mmol) was heated at 130 °C under Ar atmosphere for 40 h. The resulting solution was worked up in a manner similar to that described in the section of **3a** to give unknown product as a caramel (67.1 mg), structure of which could not determine by spectroscopic methods.

**9-(2,6-Difluorobenzyl)-1-(2-pyridinyl)hypoxanthine (8b).** A solution of **7b** (393 mg, 1.5 mmol) and copper(I) oxide (143 mg, 1 mmol) in the mixture of 2,4,6-

trimethylpyridine (5 ml) and 2-bromopyridine (0.72 ml, 7.5 mmol) was heated at 120°C under Ar atmosphere for 20 h. The resulting solution was worked up in a manner similar to that described in the section of **3a** to give pale brownish crystals (103 mg, 20%). mp 169-171°C. *Anal.* Calcd for C<sub>17</sub>H<sub>11</sub>F<sub>2</sub>N<sub>3</sub>O: C, 60.18; H, 3.27; N, 20.52. Found: C, 60.10; H, 3.35; N, 20.56. MS *m/z*: 339 (M<sup>+</sup>). UV λ max (MeOH) 256 nm. <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 8.59-8.61 (1H, m, Py H2), 8.55 (1H, s, each H2 or H8), 7.85-7.91 (3H, m, Py H4, Py H5, each H2 or H8), 7.35-7.40 (2H, m, Py H3, H4'), 6.96-7.00 (2H, m, H3', H5'), 5.47 (2H, s, -CH<sub>2</sub>-).

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